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FAQs on NCERT Books for Class 11 Chemistry 2022 The frequently asked questions regarding Chemistry 11th Class Books are given below: Q.1: Which book is best for Class 11 Chemistry? Ans: Apart from NCERT, Class 11 students can refer to the following Chemistry books for their board exam and entrance exam preparation:– Objective Chemistry (Set of 3 Vols) (S. Dinesh & Co.– Physical Chemistry By OP Tandon (G. R. Bathla Publications)– Modern's ABC Of Chemistry (Part 1 & 2)– Organic Chemistry By Himanshu Pandey (GRB Publication) NCERT books for Class 11 Chemistry have two parts, part one and two. This book contains an in-depth analysis of every topic of the curriculum prepared for CBSE. Additionally, there are excursive sections after every Chapter, which you can use for revision. You can download both the parts of NCERT Chemistry Class 11 PDF from the given link. NCERT 11 Chemistry book has a total of 14 Chapters. It includes important Chapters like the Structure of Atoms, Thermodynamics, Redox Reactions, HydroCarbons, States of Matter, Environmental Chemistry, etc. Moreover, if you are preparing for any competitive Exams like NEET, JEE, etc., this Chemistry book will become essential. Chapter 1- Some Basic Concepts of Chemistry In this Chapter, you will study the basic concept of Chemistry. Essentially, this is the introductory Chapter to the syllabus of Class 11. This Chapter began with an introduction and followed by a study of the history of this subject. After that, it moves further towards the discussion of Chemistry's importance and nature of different Matters. Under the nature of Matters section, there are two sub-sections, state and Classification. Furthermore, this Chapter discusses properties of Matters, uncertainty of measurement, laws of Chemical combination, Dalton's Atomic Theory, Atomic and Molecular Masses, etc. There are several illustrations, and an Exercise section is present at the end of this Chapter of NCERT 11 Chemistry PDF. It will help you understand the Chapter better. Chapter 2- Structure of Atom Atoms are the foundation of any Element. They are the smallest part of any Element which can Chemically react. Indian and Greek philosophers have proposed the existence of Atoms since 400 B.C. Nevertheless, the first Atomic theory was proposed by John Dalton, a British scientist, in 1808. According to his theory, Atoms were indivisible particles of Matter. However, his theory was later proved wrong towards the end of the nineteenth century. Experiments during this time proved that Atoms have three fundamental parts, Electron, proton, and neutron. This discovery of subAtomic particles gave birth to different Atomic models, which explained Atomic structures. Moreover, in this Chapter of NCERT Chemistry Class 11 PDF, you will learn about these Atomic structures, and the Exercise section will assist you to revise your learnings. Chapter 3- Classification of Elements and Periodicity in Properties Arguably, the Periodic Table is an essential concept in Chemistry. In this Chapter of NCERT books for Class 11 Chemistry, you will study about it. This unit also explains the development of Periodic law and the Periodic Table. The previous table proposed by Mendeleev was based on Atomic mass. However, the new one arranges Elements according to their Atomic number in seven horizontal rows, i.e. periods, and eighteen vertical columns that are groups or families. Along with this table, you will also study Periodic Classifications, which is a logical consequence of an Atom's Electron configuration. After that, you will Examine different Periodic trends in the Chemical and physical properties of various Elements. Chapter 4- Chemical Bonding and Molecular Structure Every Matter consists of various Elements, and under normal circumstances, no Element can exist as an independent Atom. However, noble gases are an exception. On the other hand, when a group of Atoms exists together with distinct properties, they are known as molecules. The attractive force that holds these Atoms together is called Chemical bonds. This Chapter of NCERT Class 11 Chemistry book PDF explores these bonds and structure of various Atoms. It presents an in-depth analysis of every topic within this Chapter with relevant Examples and diagrams. As a result, it is easier for you to understand this relatively complicated Chapter without much difficulty. Chapter 5- States Of Matter You have already learnt about the properties associated with a single particle. These are Atomic size, Electronic charge density, Ionization Enthalpy, Polarity, Molecular shape, etc. The characteristics of various Elements that are observable always represent bulk properties. It means that these properties are associated with a collection of a large number of Atoms, molecules, or ions. In a nutshell, you can witness characteristics of a substance in a particular state. However, you should know that the Chemical properties of substances do not change with their current state. On the other hand, their reactivity may vary. In this fifth Chapter of Class 11 NCERT Chemistry PDF, you will study the various physical Matter of substances and their characteristics. Chapter 6- Thermodynamic Molecules store Chemical energy, which they release during Chemical Reactions. This energy has a number of purposes. It can do mechanical work, can support other Reactions, etc. Hence, these different types of energies are interrelated, but only under particular circumstances. Moreover, they are transferable, from one form to another. The study of this energy transformation is known as Thermodynamics. The laws of Thermodynamics observe energy changes among macroscopic systems involving an extensive collection of molecules instead of microscopic systems with fewer molecules. In this Chapter of Thermodynamics, you will study about the terminology, heat measurement, spontaneity, etc. Moreover, take the assistance of the Exercise section at the end of this Chapter of your Class 11 Chemistry NCERT book PDF to revise this lesson. Chapter 7- Equilibrium Chemical Equilibrium is a state in which reactants are products that are both present in equations that do not tend to change with time. Hence, there are no observable changes in their properties. Typically, Equilibrium is achieved when a forward Reaction proceeds at the same rate of a reverse Reaction. However, this Reaction of backward and forward Reactions are not zero; instead, they are equal. As a result, there are no changes in the concentrations, both reactants and products. It is known as dynamic Equilibrium. In this Chapter of Chemistry 11th NCERT PDF, you will learn about it in detail. This Chapter discusses Equilibrium in both physical and Chemical processes, the law of Equilibrium and its constants, etc. Moreover, there are diagrams and equations to help you understand this concept. Chapter 8- Redox Reactions The purpose of Chemistry is to observe and present change of Matter from one kind to another. This transformation occurs via various Reactions. One important category of such Reactions is Redox Reactions. In the case of Redox Reactions, Oxidation and Reduction coincide. Moreover, Redox Reactions have four categories, combination, decomposition, displacement, and disproportionation. Along with its types, Redox couple and electrode processes have been discussed in this Chapter as well. Redox Reactions have a wide range of applications in pharmaceutical, agricultural, metallurgical, etc. areas. Furthermore, the equations and problems attached to this Chapter of NCERT Chemistry Class 11 PDF will help you understand this topic better. Chapter 9- Hydrogen Compared to the other Elements on the Periodic Table as well as nature, Hydrogen has the simplest Atomic structure. In its Atomic form, it has one proton and one Electron. On the other hand, in its Elemental form, it is a Diatomic molecule known as Dihydrogen. Hydrogen forms more compounds in nature than any other Element. It is also the first Element on the Periodic Table. Additionally, Hydrogen has great industrial importance as well. In this Chapter of NCERT 11 Chemistry book, you will learn about Hydrogen, Dihydrogen, its properties, preparations, hydrides, etc. Additionally, solve the Exercise section to know more about this Element. Chapter 10- The s-Block Elements s-Block Elements are those in which the last Electron enters their furthest s-orbital. Since this outermost s-orbital can house only two Electrons, groups one and two belong to this section of the Periodic Table. The first group consists of Elements like lithium, sodium, rubidium, potassium, caesium, and francium. These Elements are popular alkali metals. On the other hand, the second group comprises Elements such as beryllium, calcium, magnesium, strontium, radium, and barium. These Elements, except beryllium, are popularly known as the alkaline earth metals. Moreover, in this Chapter of NCERT 11th Chemistry book PDF, you will learn about these Elements in detail. You will study their properties and usage as well. Additionally, completing the Exercise will allow you to understand these s-Block Elements even better. Chapter 11- The p-Block Elements In the case of p-Block Elements, the last Electron enters the furthest p-orbital. Since there are three p-orbital, a set of six Electrons can be accommodated in one p-orbital. Consequently, there are a total of six such sets of Elements in the entire Periodic Table. It ranges from number 13 to 18. Boron, Nitrogen, Carbon, Oxygen, Helium, Fluorine head these groups. On the other hand, the inner core of Electron configuration may differ. This difference influences the physical and Chemical properties of an Element. Therefore, you can observe variations in terms of the properties of these p-Block Elements. Chapter 11 of NCERT Chemistry Class 11 PDF explains this set of Elements even further. You can learn their characteristics and their usage as well from this Chapter. Chapter 12- Organic Chemistry – Some Basic Principles and Techniques In the previous Chapter, i.e. Chapter 11, you studied how Carbon has a distinctive property named catenation. Due to this property, it creates covalent bonds with other Carbon Atoms. Moreover, it can also form the same Bonding with Atoms of different Elements like Oxygen, Hydrogen, Sulphur, Nitrogen, Halogens and Phosphorus as well. These new Elements and their properties are studied under a separate branch of Chemistry called Organic Chemistry. In this Chapter, you will learn about its fundamental principles, its technique of analysis, its formation, etc. Additionally, these new Elements are known as Organic compounds. There are several diagrams in this Chapter of 11th Chemistry NCERT PDF, which will help you understand the Chemical structure of these compounds. Chapter 13- Hydrocarbons Hydrocarbons represents a compound that has Carbon and Hydrogen molecules. Coal and petroleum Elements are some of the most common hydroCarbon compounds. Besides, hydroCarbons are Classified into three groups. These are -Open chain saturated (alkanes) Unsaturated (alkenes and alkynes) Cyclic (alicyclic) and aromatic In this Chapter of NCERT Chemistry Class 11 PDF, you will learn about the Classification of hydroCarbons, alkanes, alkynes, and aromatic hydroCarbons. Additionally, you will study these groups in detail, their properties, nomenclature, isomer, etc. are included here. Chapter 14- Environmental Chemistry In Environmental Science, you have already studied the economic, social, Chemical, biological, and Chemical relation with the surroundings. However, in this Chapter of 11th Chemistry NCERT PDF, the focus is solely on Chemistry. In Environmental Chemistry, you will learn about the origin, transport, Reactions, effects and fate of various Chemical species in the Environment. Moreover, in this Chapter, you will study the following topics – Environmental Pollution Atmospheric Pollution Water Pollution Soil Pollution Industrial Waste Strategies to Control Environmental Pollution Green Chemistry This Chapter will provide you with a more in-depth idea of various problems associated with the Environment. Apart from that, there are numerous short trivia mentioned in this Chapter, as well. They will help you understand any topic better. Advice from experts that can help students top the CBSE Chemistry Exam for Class 11: Chemistry, like other math-heavy Sciences, is a topic that takes a lot of time and effort to master. Chemistry involves memorizing computations, formulas, and diagrams, while others necessitate a thorough comprehension of Chemical structures and math calculations. In Class, pay attention to your Chemistry teacher. Because the instructor is present to clarify and explain topics, this should make learning easier. Don't take this time for granted; instead, concentrate on understanding everything your teacher says. If you have any questions or concerns about any points, equations, or themes, don't hesitate to bring them up in Class. If you're having trouble with the Chemical topic, talk to your teacher or Classmates about it. Ask your instructor for study guides or worksheets if you're having trouble grasping a concept. It's ideal to go over your notes as soon as possible after Class; if you wait until the weekend, you might forget some of the material. Reading from your own notes rather than a textbook may make it simpler to remember things since you will have previously written them in words you understand. You'll be more grounded and ready to grasp specifics within the area if you know what broad topic of Chemistry you're studying and what it entails. The following are the five categories: The study of the structures and compounds of substances generated from all Elements except Carbon is known as inorganic Chemistry. Organic Chemistry is all about the study of Carbon compounds and Chemical bonds. The use of mathematical approaches to Chemical issues is known as physical (or theoretical) Chemistry. Analytical Chemistry Examines Chemical substances quantitatively and qualitatively. The study of Chemical interactions within live cells and the Chemical structures of living Matter is known as bioChemistry (or physiological Chemistry). To get good grades and score well in Examinations, take the above assistance and put it into practice. Follow NCERT books to gain a thorough understanding of concepts and solve numerical problems to become accustomed to the difficulties. Students, as well as Teachers, Prefer NCERT Books NCERT books are popular among students and teachers due to several reasons. 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Another significant advantage of these NCERT Chemistry Class 11 PDF is that it put special emphasis in clearing the basic concepts of Chemistry. Along with its languages and illustration, these books aim to simplify the complicated Chapters of Chemistry. Comprehensive Study Material These books are comprehensive in terms of the information. Typically, extensive research goes behind preparing these books. Hence, they cover every aspect of a topic. The Exercise section in these books is very useful. They comprise questions which are designed to enhance a student's problem-solving capability. Choose Vedantu for Better Preparation Vedantu is one of the leading e-learning platforms of India. With the help of their website and app, students can enhance their CBSE Exam preparations. Students can find essential Exam notes, study materials, practice sets, etc. here. Moreover, they can also find solutions to their NCERT books on the app and website of Vedantu. 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